NANOSTRUCTURAL THERMOELECTRIC MATERIALS OBTAINED BY SOLVOTHERMAL SYNTHESIS AND HOT ISOSTATIC PRESSURE

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1 Introduction

Thermoelectric materials are of interest for applications in electrical power generation devices and solid-state cooling due to many attractive properties (long life, no emissions of toxic gases no moving parts, low maintenance, etc).

At present bismuth telluride based compounds are known to be the most excellent thermoelectric materials for around room temperature applications.

The Bi₂Te₃-based alloys are acceptable for some specialized applications, but they are far less so for commercial refrigeration on a large scale. A number of investigations have focused on optimizing the composition, tuning doping with other heavy metals, optimizing device design, etc in order to improve thermoelectric properties of the Bi₂Te₃-based materials. However, the thermoelectric efficiency of these materials has not improved obviously and the dimensionless figure of merit (*ZT*) has been approximately 1 for many years.

According to theoretical and experimental investigations, the thermoelectric nanomaterials, such as quantum wells, superlattice, quantum wires, nanograined thin films, bulk nanocomposites demonstrate much higher thermoelectric coefficients than their traditional alternatives [1-4].

Bulk nanostructured materials are now considered as one of perspective thermoelectric materials. A specific technology should be developed to fabricate nanostructured thermoelectric materials with reproducible and advanced properties.

One of technological approach is based on two principal stages as follows [5]:

- Synthesis of initial nanopowder with desired structure, phase and element compositions, size and shapes of particles, etc.
- Consolidation of synthesized nanopowder by using a pressing and high temperature treatment in order to retain a nanostructure and fabricate dense sample with high enough mechanical strength and thermoelectric parameters.

In present work such a kind of technology based on microwave-solvothermal synthesis and hot isostatic pressing was applied to prepare the bulk nanostructured Bi₂Te₃-based material. As is known, compared with the conventional methods, the microwave-assisted heating technique has the advantages of very short time, simplicity and energy efficiency, small particle size of the products, narrow particle size distribution and high purity [6].

2 Experimental procedure

Bismuth telluride nanopowders have been prepared via microwave-solvothermal synthesis in closed reactor ERTEC (Model 02-02).

Analytical grade Bi₂O₃, TeO₂ and ethylene glycol were used as starting components. A 110 mL teflon-lined stainless-steel autoclave was used and the temperature was regulated by a digital-type temperature-controlled oven. Microwave assisted reactions were conducted in a 300 W microwave oven with a 2450 kHz working frequency.

The ethylene glycol was used as both the solvent and the reducing agent in the reaction. A few routes of synthesis were applied to determine optimal reaction conditions (Table 1). After synthesis, the reaction product as a black precipitate was washed with alcohol and then centrifuged and dried.

Morphology and structure of synthesized powder were characterized by X-ray diffraction (XRD) using a Rigaku Ultima IV diffractometer with $\text{Cu}K_{\alpha}$ -radiation, transmission electron microscopy (TEM) using a JEM-2010 microscope and scanning electron microscope (SEM) using a Zeiss LEO 1530 microscope.

Synthesized nanopowders were hot isostatically pressed (HIP) at temperature of 400 °C during 5 min by using a toroidal press. Powder for compaction was placed in graphite matrix with hexagonal BN powder as media to spread isostatic pressure to the object under pressing. Pressures at 2, 4, 6 and 8 GPa were used. Microstructure of consolidated material was

then investigated by XRD and SEM using a Zeiss LEO 1530 scanning electron microscope. EDAX (Energy Dispersive X-ray Microanalysis) method was used to study an element distribution within the samples under consolidation.

Electrical conductivity, σ , of the consolidated samples was also measured by four-probed method at room temperature.

3 Results and discussion

It was established that phase composition of material after synthesis is strongly dependent on synthesis conditions and ratio of initial reagents. Phase compositions of the powder samples synthesized at various conditions are collected in Table 1.

One can see that all of five technological routes allow us to prepare a desired Bi₂Te₃ phase, but for routes No 1, 2, 3 and 5 the Bi₂Te₃ phase is coexisting with other parasite phases (BiTe, Bi₄Te₃, Bi, Te). Single phase Bi₂Te₃ powder could be prepared via route No 4. This powder synthesized at optimal conditions was used for further study.

Morphology of the $\mathrm{Bi}_2\mathrm{Te}_3$ powder has been investigated by SEM (Fig. 1). It is established that powder after synthesis consists of agglomerate of particles with average size of 200 nm. TEM-image in Fig. 2 shows typical morphology of microwave-solvothermally synthesized powder. It is seen that powder contains plate-like nanoparticles with average size about 30 nm.

Bulk material cylindrical form with sizes of 5x5 mm was then prepared by HIP-consolidation.

A few specific features were found at research of consolidated material:

- In contrast with initial powder, the phase composition extracted from XRD patterns showed presence of Bi₂Te₃ (space symmetry group *R*-3*m*) and BiTe (*P*-3*m*1). So, at high temperature and under high pressure some part of the Bi₂Te₃ phase transforms into the BiTe phase.
- Consolidated materials have dense, homogeneous and porousless nanocrystalline structures (Fig. 3). Grain size, *d*, of the material under study is a HIP-pressure-dependent and average grain size is changed from 60 to 100 nm (Fig. 4).
- EDAX experiments confirmed a homogeneous distribution of the Bi and Te elements within the material.

According to theoretical consideration [7], electrical conductivity of bulk thermoelectric Bi₂Te₃ material should be depressed when grain size is decreasing. This behavior is attributed to carriers mobility decrease owing a carriers scattering by grain boundaries. The $\sigma(d)$ dependence is shown in Fig. 5. For the grain sizes of 100, 85 and 60 nm (corresponding HIP-pressures are 2, 6 and 8 GPa) experimental points are in agreement with theoretical prediction. But the electrical conductivity is maximum for the sample with d=80 nm (P=4 GPa). It is obviously that to explain the change of electrical conductivity versus the HIP-pressure other physical mechanisms besides grain size change should be taken into account. In particularly, carriers concentration and defect structure in volume materials can also change during the hot isostatic pressure.

Characterization of the thermoelectric properties of the Bi₂Te₃-based nanostructured materials is in progress now.

4 Conclusion

Single-phases Bi₂Te₃ plate-like crystals with homogeneous hexagonal morphology were rapidly synthesized using by the microwave assisted solvothermal method in 50 min at 250°C and 30 atm. Synthesized nanopowder consists of particles agglomerate with average size of 200 nm, which consists of crystals with size about 30 nm. HIP compaction of powders at temperature of 400°C and pressures of 2, 4, 6 and 8 GPa formed homogeneous Bi₂Te₃-based material with average grain size is of 60-100 nm. Electrical conductivity of the samples under study shows a complex dependence of the grain size (HIP-pressure).

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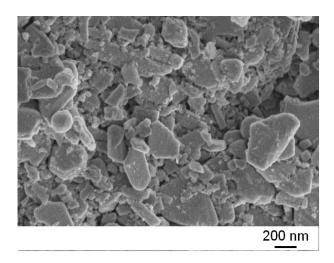


Fig1. Morphology of the Bi₂Te₃ nanopowder by SEM

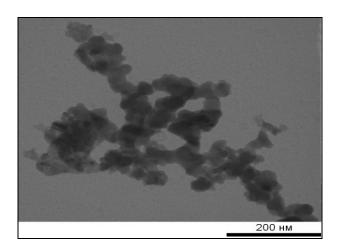


Fig.2. Nanocrystals of the Bi₂Te₃ powder by TEM

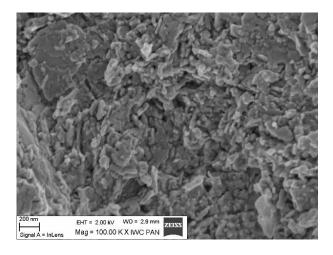


Fig. 3. Microstructure of the Bi_2Te_3 –based materials consolidated by HIP method at temperature of $400^{\circ}C$ and pressures of 8 GPa

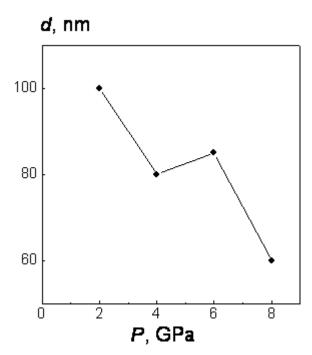


Fig.4. The d(P) dependence for the Bi₂Te₃ –based materials consolidated by HIP method

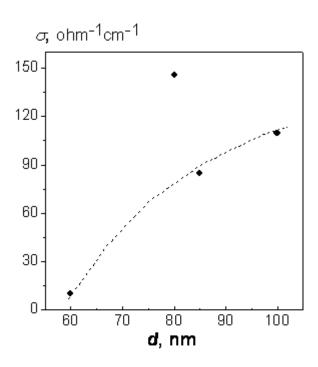


Fig. 5. The $\sigma(d)$ dependence for the Bi₂Te₃ –based materials consolidated by HIP method

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